

### CLAIMS

The embodiment of the invention in which an exclusive property or privilege is claimed is defined as follows:

- 1        1. An improved electrolyte for the electrolysis of alumina ( $\text{Al}_2\text{O}_3$ ), the electrolyte  
2        comprising a mixture of aluminum fluoride ( $\text{AlF}_3$ ) and potassium fluoride (KF).
- 1        2. The electrolyte recited in claim 1 wherein the aluminum fluoride ( $\text{AlF}_3$ ) is present  
2        in a molar ratio with potassium fluoride (KF) greater than 1:1.
- 1        3. The electrolyte recited in claim 1, the electrolyte further comprising from about 2  
2        to 6 wt. % of aluminum oxide (alumina/ $\text{Al}_2\text{O}_3$ ).
- 1        4. The electrolyte as recited in claim 1 wherein the concentrations of the  
2        electrolyte's components remain constant during electrolysis.
- 1        5. The electrolyte as recited in claim 1 where the electrolyte is used to  
2        electrolyze the alumina to aluminum at from about  $660^\circ\text{C}$  to  $1000^\circ\text{C}$ .
- 1        6. The electrolyte as recited in claim 3 wherein the electrolyte remains liquid during  
2        electrolysis.
- 1        7. The electrolyte as recited in claim 6 wherein inert anodes and wettable  
2        cathodes can be used with the electrolyte.

1           8.     A method for using inert anodes and wettable cathodes in the electrolytic  
2 production of aluminum comprising using the anodes and cathodes in an electrolysis  
3 process at a temperature from between 660°C and 1000°C.

1           9.     The method as recited in claim 8 wherein the anode is comprised of a  
2 material selected from the group consisting of metals, metal alloys, metal oxides, and  
3 cermets.

1           10.    The method as recited in claim 8 wherein the alloys are copper alloys  
2 (aluminum-bronze) and are from about 1 wt.% to 20 wt.% aluminum (Al).

1           11.    The method as recited in claim 8 wherein the cathode is made of a material  
2 selected from the group consisting of metal borides, metal boride-carbon composites,  
3 metal boride-containing coatings on substrates, graphite, molybdenum, and tungsten.

1           12.    The method as recited in claim 8 where the anode contains an oxygen-  
2 containing film.

1           13.    The method as recited in claim 12 wherein the oxygen containing film is  
2 formed during the electrolysis process.

1           14.    The method as recited in claim 11 wherein the material coats one side of  
2 the inert anode to create a bipolar electrode for use in multipolar electrolytic cells.

1           15.    The method as recited in claim 8 wherein the distance between the anode  
2 and cathode remains constant.

1           16.    The method as recited in claim 8 wherein the electrodes can be arranged  
2 in a vertical configuration.

1           17. The method as recited in claim 8 wherein the electrodes can be arranged  
2 in a slanted configuration.

1           18. The method as recited in claim 8 wherein O<sub>2</sub> is the only gas generated by  
2 electrolysis.

1           19. The method as recited in claim 16 wherein the vertical configuration of the  
2 cell creates electrolyte recirculation pathways.

1           20. A method for electrolyzing alumina below 1000°C, the method  
2 comprising:

- 3           a) supplying an electrolyte containing more than 35 mol% potassium  
4 fluoride (KF) and more than 30 mol% aluminum fluoride (AlF<sub>3</sub>);  
5           b) injecting alumina with the electrolyte; and  
6           c) subjecting the electrolyte to a voltage.